Enantioselective Formal Synthesis of (−)-Podophyllotoxin from (2S,3R)‑3-Arylaziridine-2-carboxylate

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S Supporting Information

ABSTRACT: Meyers' 4-aryl-1-tetralone-lactone and ent-Zhang's 2-diarylmethyl-4-oxobutanoate were synthesized in the formal synthesis of (−)-podophyllotoxin from (2S,3R)-3-arylaziridine-2-carboxylate, via 3,3-diarylpropanoate as a common intermediate, in an overall 42% yield through 10 steps and 31% yield through 6 steps, respectively. The key steps in the synthesis were regioand diastereoselective ring opening with an aromatic nucleophile, samarium iodide promoted reductive C−N bond cleavage, and Stille coupling for introducing the vinyl functionality. The starting aziridine was enantioselectively prepared from 3,4, 5-trimethoxybenzaldehyde by guanidinium ylide mediated asymmetric aziridination. All nitrogen components used in the reaction sequence are reusable as the starting guanidinium source.

■ INTRODUCTION

Lignans are a family of compounds consisting of dimerized C6−C3 phenylpropanoids, which are biosynthetically derived from shikimic acid via an aromatic α -amino acid such as phenylalanine or tyrosine. Among lignans, the aryltetralin-lactone (−)-podophyllotoxin (1) is an important natural product because of its potent antitumor activity; in fact, sugar-modified analogues, such as etoposide and etopophos, are clinically used as anticancer drugs.¹ In the stereoselective synthesis of 1 and its analogues, the challenges are the four contiguous stereocenters and the prese[nc](#page-10-0)e of a base-sensitive trans-lactone moiety. Although many approaches² to the racemic synthesis of podophyllotoxin can be found in the literature, to the best of our knowledge, only eight groups³ ha[ve](#page-10-0) reported its enantioselective total synthesis. Meyers and co-workers^{3g} used conjugate addition of $3,4,5$ -trimethoxyphen[yl](#page-10-0) anion to a chiral 2-naphthalenyloxazoline to construct an aryltetralin ske[let](#page-10-0)on in the asymmetric induction step and succeeded in the first total synthesis of natural (-)-podophyllotoxin (1) through an aryltetralone-lactone, (−)-3-(hydroxymethyl) picropodophyllone (2) . Recently, Zhang and co-workers^{3a} completed a concise total synthesis of $ent-(+)$ -podophyllotoxin (ent-1) by applying a tandem Michael addition−allylation of [a](#page-10-0) chirally masked 2-piperonyl anion to cinnamate and allyl bromide as an asymmetric induction step, in which they used 2-diarylmethyl-4-oxobutanoate 3 as a key precursor for constructing the aryltetralin skeleton.

We⁴ have previously developed an atom-economical method, applicable to asymmetric synthesis, for preparing aziridine-2-car[bo](#page-10-0)xylates by reaction of a guanidinium salt with an aromatic (or unsaturated) aldehyde: a trans-aziridine is formed as the major product when an electron-rich aldehyde is used as the electrophile, and the coformed urea can be reused to prepare the starting guanidinium salt. We^{4c,e,5} also examined the ring opening of the generated 3-aryl (or unsaturated) aziridine-2-carboxylates by various nucleophiles, whi[ch pr](#page-10-0)ovided the nucleophile-inserted 2-amino-3-aryl (or unsaturated) propanoates. The 2-amino-3, 3-diarylpropanoate derivative formed by the reaction with an aromatic (Ar) nucleophile is a suitable synthetic precursor for 4-aryltetralin-type lignans.

In this paper we disclose new enantioselective syntheses of Meyers' 4-aryl-1-tetralone-lactone (2) and ent-Zhang's 2-diarylmethyl-4-oxobutanoate (3), which are useful synthetic intermediates for (−)-podophyllotoxin (1), from trans-(2S,3R)-3-arylaziridine-2 carboxylate through the ring-opened 2-amino-3-arylpropanoate. The key steps in the synthesis are guanidinium ylide mediated asymmetric aziridination of 3,4,5-trimethoxybenzaldehyde, regio- and diastereoselective ring opening of the generated trans-(2S,3R)-3-(3,4, 5-trimethoxyphenyl)aziridine-2-carboxylate with sesamol (3,4-methylenedioxyphenol) as an Ar nucleophile, samarium iodide (SmI₂) promoted reductive C−N bond cleavage of the ring-opened 2-amino-3,3-diarylpropanoate derivative, and Stille coupling for introducing a vinyl functionality to the 2-hydroxy-4,5-(methylenedioxy)phenyl substituent. In addition, each nitrogen component generated in the aziridination and in the C−N bond cleavage reaction could be recovered and recycled to prepare the guanidinium starting source.

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Scheme 1. Retrosynthetic Analysis

■ RESULTS AND DISCUSSION

We observed that nonactivated N-benzyl-3-arylaziridine-2-carboxylates were attacked exclusively at the C3 benzylic position by electron-rich Ar nucleophiles in the presence of indium chloride $(InCl₃)$ as a Lewis acid (LA) catalyst to give 2-benzylamino-3,3-diarylpropanoates with high diastereoselectivity.5b X-ray crystallographic analysis of a ring-opened product established that the stereochemistry of an Ar-inserted C3 ster[eo](#page-10-0)genic center was controlled by an anti-mode nucleophilic opening process (inversion). Against this background, our retrosynthetic analysis of Meyers' 4-aryl-1-tetralone-lactone (2) and ent-Zhang's 2-diarylmethyl-4-oxobutanoate (3) is shown in Scheme 1. trans-(2S,3R)-N-Benzyl-3-(3,4,5-trimethoxyphenyl) aziridine-2-carboxylate trans- $(2S,3R)$ -6, prepared by an asymmetric aziridination reaction^{4,5} of 3,4,5-trimethoxybenzaldehyde (7) using the (R,R) -guanidinium salt 8, is converted to $(2S,3R)$ -2-benzylamino-3,3-diarylpropa[no](#page-10-0)ate 5 by regio- and diastereoselective ring opening with a 4-substituted methylenedioxybenzene. Reductive cleavage of the C−N bond in the aminopropanoate 5 provides the propanoate 4 and benzylamine, either before or after vinylation

through a palladium (Pd)-catalyzed coupling reaction. The benzylamine obtained could be reused as the external nitrogen source in the preparation of a starting guanidinium salt 8 by reaction with urea 9, coproduced in the aziridination reaction, after chlorination. The propanoate 4 could then be converted to the targeted 2 and 3 by suitable chemical manipulations.

In our synthetic scheme, the key step is the stereoselective construction of the C3 stereogenic center in 2-amino-3,3 diarylpropanoates 5 by ring opening of trans-(2S,3R)-3-(3,4,5 trimethoxyphenyl)aziridine-2-carboxylate trans-(2S,3R)-6. The starting trans-(2S,3R)-6 was prepared in 84% yield and with 82% enantiomeric excess (ee) by asymmetric aziridination using the (R,R) -guanidinium salt 8 under one of our standard conditions using tetramethylguanidine (TMG) as base, 4.5 in which the diastereoisomeric cis-(2S,3S)-6 and urea 9 were coproduced in 4% (84% ee) and 83% yields, respectively. In addition, t[he c](#page-10-0)orresponding racemic aziridines (\pm) -trans- and (\pm) -cis-6 were also prepared by using guanidinium salt 10, which lacks phenyl pendants, under an alternative standard condition (NaH in DMF)^{4,5} as a model source for ring-opening experiments (Scheme 2).

Scheme 2. Aziridination

Scheme 3. Preliminary Examination of InCl₃-Catalyzed Ring-Opening Reactions

At first, we preliminarily examined the ring-opening reaction using the racemic *trans*-aziridine (\pm) -trans-6 with sesamol derivatives (Scheme 3a). Treatment with sesamol benzyl ether in the presence of a 0.5 mol amount of InCl₃ afforded the desired carbon-inserted ring-opened product (\pm) -11a $(R = Bn)$ in 55% yield with a moderate diastereomeric ratio $(dr = 5.6:1)$. Similar results were obtained with an allyl ether, even when the amount of InCl₃ was reduced (0.5 equiv \rightarrow 0.3 equiv) ((\pm)-11b: 60% yield, dr = 4.3:1). Gratifyingly, the use of sesamol led to satisfactory conversion, even with the lower amount (0.1 equiv) of $InCl₃$, in which the ring-opened 2-amino-3,3-diarylpropanoate (\pm) -11c (R = H) was provided in 91% yield, but the diastereoselectivity was slightly lowered $(dr = 4.0:1)$.

An alternative approach involving interchange of the aziridine unit from the trimethoxyphenyl- (\pm) -trans-6 with the methylenedioxyphenylaziridine (\pm) -trans-12^{4e,f,5} and the Ar nucleophile unit from sesamol with the trimethoxybenzene derivative, respectively, may also be possible, despite the undesired formation of the diastereomeric ring-opened product 13 ($X =$ OMe) with an opposite stereochemistry at the C3 stereogenic center. It had been reported that the 2,3,4-trimethoxyphenylinserted product, not the corresponding 3,4,5-trimethoxyphenyl derivative, was provided in the ring opening of an activated N -tosylaziridine derivative with 1,2,3-trimethoxybenzene.⁶ Furthermore, we experienced unsuccessful ring opening when anisole was used as an Ar nucleop[h](#page-10-0)ile.^{5b} Thus, more electron-rich 2,6-dimethoxyphenol and 2,6-dimethoxyaniline were selected as potential 3,4,5-trimethoxyphenyl-t[ype](#page-10-0) nucleophiles and we comparably attempted the InCl₃-catalyzed ring opening of the methylenedioxyphenylaziridine (\pm) -trans-12 (Scheme 3b). However, different from the sesamol-type nucleophiles, they functioned as heteroatom nucleophiles, not as Ar nucleophiles, to give the

heteroatom-inserted 2-aminopropanoate (\pm) -14 as a single diastereoisomer, in place of carbon-inserted 2-amino-3,3-diarylpropanoates 13. In particular, the product (\pm) -14b $(X = NH)$ was provided in high yield (92%) when 2,6-dimethoxyaniline was used as a nucleophile. These facts may suggest that a softer C6 atom of sesamol, doubly activated by two electron-donating groups at 1,3-positions, in comparison to the phenolic oxygen atom serves as the nucleophilic center in the ring-opening reaction of an aziridine substrate through C−C bond formation.⁷

The result regarding ring opening of trimethoxyphenylaziridine (\pm) -trans-6 with sesamol prompted us to optimize an L[A](#page-10-0) catalyst (Table 1). Although conversion was slightly increased by the replacement of InCl₃ with indium triflate $(In(OTf)_{3})$ or scandium triflate $(Sc(OTf)_3)$, no improvement in dr was observed (Table 1, entries 1−3). High dr (10:1) was attained

Table 1. Screening of LAs for the Ring Opening of the Trimethoxyphenylaziridine (\pm) -trans-6 with Sesamol

with zinc triflate $[Zn(OTf)_2]$ or copper triflate $[Cu(OTf)_2]$, and in the former case, the high conversion (89%) was maintained (Table 1, entries 4 and 5). On the other hand, boron trifluoride etherate $(BF_3·Et_2O)$ or zinc iodide (ZnI_2) led to unsatisfactory results (Table 1, entries 6 and 7). Thus, we selected $Zn(OTf)₂$ as the most suitable LA catalyst for the sesamol-participated ring opening and obtained the same results when the enantiorich aziridine $trans-(2S,3R)$ -6 was applied to the ring-opening reaction.

Next, we attempted to introduce a vinyl unit into the sesamolinserted 2-amino-3-arylpropanoate 11c, derived from trans- (2S,3R)-6, by Pd-catalyzed coupling before reductive C−N cleavage (Scheme 4). After the phenolic function in 11c was activated by making trifluoromethanesulfonate, the N-benzyl $(N-Bn)$ derivative [1](#page-4-0)5 was subjected to a Stille-type reaction.⁸ Coupling product 16 was afforded in moderate yield; however, subsequent reductive C−N cleavage of 16 under the optimize[d](#page-10-0) conditions discussed later was unsatisfactory (4: 30%, see Scheme 5).

Independent trials for introduction of a C3 unit under Heck⁹ or Sonogashira¹⁰ reaction conditions as alternative approache[s t](#page-5-0)o the Meyers intermediate (2) were attempted. However, startin[g](#page-10-0) materials were [m](#page-10-0)ainly recovered in both coupling reactions using either the N-Bn (15) or N-(tert-butoxycarbonyl) (N-Boc) derivative (17), which was derived from 15 by catalytic hydrogenation in the presence of di-tert-butyl dicarbonate $((Boc)₂O)$. Therefore, 2-aminopropanoate 11c was subjected to a reductive C−N bond cleavage reaction before introduction of the C unit.

Honda and Ishikawa¹¹ reported the SmI₂-promoted C−N bond cleavage of α -amino carbonyl compounds containing amino esters. In this rea[cti](#page-11-0)on, a combination of tetrahydrofuran (THF), hexamethylphosphoric triamide (HMPA), and methanol (MeOH), serving as a solvent, additive, and proton source, respectively, played an important role in reaction completion. However, we observed no reaction when 2-aminopropanoate 11c was subjected to Honda and Ishikawa's condition using MeOH as a proton source. Successful C−N bond cleavage was accomplished by replacing MeOH with water, which provided propanoate 18 and benzylamine in the same 93% yield (Scheme 5). The vinyl-inserted product 4 was smoothly provided by a Pd-coupling reaction of 18 under Stille-type conditions,⁸ after conversion [t](#page-5-0)o trifluoromethanesulfonate 19, in better yield (90% at 50 $^{\circ}$ C for 7 h), in comparison with the case of 2-aminopropan[o](#page-10-0)ate 15 (64% at 90 °C for 8 h; Scheme 4). The ee of 4 was estimated as 81%, without the loss of enantiopurity through reactions, and the ee could be increased to 99% after recry[sta](#page-4-0)llization from hexane/MeOH (10/1). This fact is relevant to our synthetic strategy, because vinyl product 4 was the common key precursor for Meyer's (2) and ent-Zhang's intermediates (3). In trials for introduction of a C3 unit into the deaminated product 19 under Heck reaction conditions,⁹ different from the case of 2-aminopropanoates 15 and 17 (Scheme 4), the coupling product 20 was obtained, even in moderate yield[. H](#page-10-0)owever, reduction of the acrylate in 20 to the allyl alcohol failed un[de](#page-4-0)r the conditions examined.

On the basis of these experimental facts, chemical manipulation of 4 toward the Meyers intermediate (2) was examined. Straightforward preparation of tetralinol (23) and/or tetralone systems (26) were initially attempted (Scheme 6). Direct epoxidation of 4 under conventional conditions failed, but the intended epoxide 21 was afforded in good yield by [a](#page-6-0) stepwise approach (93% in two steps) through iodohydrin 22, which was provided in 95% yield by treatment of 4 with N-iodosuccinimide (NIS) in aqueous THF. Although the diastereoselectivity in the products was low $(dr = 2.1)$, the newly created benzylic stereogenic center will be destroyed in a later conversion to a ketone. Unfortunately, cyclization of epoxide 21 or iodohydrin 22 to tetralinol 23 was unsuccessful. Instead, indane derivative 24 displaced at the benzylic position was obtained in approximately 30% yield. Trial reactions for constructing the tetralone system 26 from iodo ketone 25, which was available as an unstable product from iodohydrin 22 by Dess−Martin periodinane (DMP) oxidation, also failed.

Therefore, we applied a stepwise approach to the tetralin systems (Scheme 7). After protection of the alcohol group of 22 with a tert-butyldimethylsilyl (TBS) group, treatment of the silyl ether 27 with lit[hi](#page-6-0)um hexamethyldisilazide (LHMDS) in the presence of HMPA afforded the desired cyclized tetralin 28 in 90% yield as expected. The $^1\mathrm{H}$ NMR spectrum showed that the 2:1 dr of starting material 22 was preserved in product 28, indicating that highly stereocontrolled ring closure occurred; however, the relative stereochemistry between C3 and C4 configurations (numbering based on tetralin-1-ol) could not be assigned because of complex signal patterns in the ¹H NMR spectrum. After deprotection, oxidation of tetralinol 23 with DMP provided the corresponding ketone 26 as a single isomer in 92% yield. The desired cyclization from protected iodohydrin 27 to tetralinol system 28 with a *trans*- $(3R,4R)$ configuration was deduced by comparison with the coupling constant between C3–H and C4–H ($J = 8.0$ Hz) in tetralone system 26.¹² Aldol

condensation of 26 with excess formalin solution smoothly provided (−)-3-(hydroxymethyl)picropodophyllone (2), Meyers' intermediate, in 95% yield. Product data were identical with reported data, $3g$ and the overall yield was 35% from a commercially available 3,4,5-trimethoxybenzaldehyde (7) through 11 steps with an aver[age](#page-10-0) 91% yield. On the other hand, Meyers and co-workers synthesized the intermediate 2 in overall 10% yield from piperonal through 19 steps with an average yield of 89%.

Finally, synthesis of ent -Zhang's intermediate (3) from a common precursor (vinyl-inserted product 4) was attempted (Scheme 8). Alkylation with allyl bromide provided 2-diarylmethyl-4-pentenoate 29 in 92% yield with a 3:1 dr. Oxidative cleavage of two vi[ny](#page-7-0)l functions with an osmium tetraoxide $(OsO₄)$ and sodium periodate (NaIO4) system proceeded smoothly to give the desired 2-diarylmethyl-4-oxobutanoate (3), ent-Zhang's intermediate, in 49% isolated yield after purification of the diastereoisomeric dicarbonyl compounds obtained. These findings indicate that the major diastereoisomer of the pentenoate 29 was the 2S derivative. The data on the product 3 were identical with the reported data, except for a specific rotation, $3a$ and the overall yield was 26% from 3,4,5-trimethoxybenzaldehyde (7) through seven steps with an average yield of 84%. T[he](#page-10-0) synthetic efficiency was almost comparable to that of Zhang's route (overall 34% yield from bromopiperonal through four steps with an average yield of 78%).

■ CONCLUSION

We effectively achieved the enantioselective syntheses of Meyers' 4-aryl-1-tetralone-lactone and ent-Zhang's 2-diarylmethyl-4 oxobutanoate as useful synthetic intermediates for (−)-podophyllotoxin from trans-(2S,3R)-3-(3,4,5-trimethoxyphenyl)aziridine-2-carboxylate, which was derived from 3,4,5-trimethoxybenzaldehyde by guanidinium ylide mediated asymmetric aziridination. This strategy may appear to have low atom economy because nitrogen

atoms are not incorporated into the products, but the nitrogen components (urea and benzylamine) released during the reaction sequence can be recovered and reused to prepare the guanidinium starting material. In addition, not only is the aziridine system likely analogous to an aromatic α -amino acid derivative-a biosynthetic precursor of phenylpropanoid as a monomeric lignan component-but it might also contribute greatly to controlling the new stereogenic center of the diarylsubstituted carbon in the ring-opened product. Thus, this approach provides a new strategy for the synthesis of biologically important 4-aryltetralin lignans.

EXPERIMENTAL SECTION

Melting points were determined with a melting point hot-stage instrument and are uncorrected. ${}^{1}H$ and ${}^{13}C$ NMR spectra were recorded with 400 and 600 MHz spectrometers in CDCl₃. FAB and ESI mass spectra were obtained using a double-focusing magnetic sector mass spectrometer and time-of-flight mass spectrometer, respectively. For column and flash chromatography silica gel 60 (63-210 μ m) and NH-coated silica gel (100−200 mesh) and silica gel 60 (40−100 $μ$ m), respectively, were used. All reactions were carried out using dry solvents under an argon atmosphere, and organic extracts were evaporated under reduced pressure after drying over MgSO₄, unless otherwise noted.

Asymmetric Aziridination of 3,4,5-Trimethoxybenzaldehyde (7) and (R,R)-Guanidinium Bromide (8). A freshly distilled TMG (0.4 mL, 3.20 mmol) was added dropwise to a stirred solution of 7 (0.589 g, 3.00 mmol) and 8 (1.732 g, 3.146 mmol) in THF (3.0 mL) at 0 °C. The mixture was stirred at room temperature for 1 day, and then SiO_2 (30 g) was added after dilution with CHCl₃ (30 mL). The resulting slurry was stirred at room temperature for 1 day. The $SiO₂$ was filtered through Celite and washed with AcOEt. After evaporation of the filtrate, column chromatography of the residue ($SiO₂$, AcOEt/hexane 1/6) gave trans-(2S,3R)-6 (1.01 g, 84%, 82% ee), cis-(2S,3S)-6 (43 mg, 4%, 84% ee), and urea 9 (colorless prisms (692 mg, 83%), mp 149−151 °C).

trans-(2S,3R)-tert-Butyl 3-(3,4,5-trimethoxyphenyl)aziridine-2 carboxylate (trans-(2S,3R)-6): colorless needles, mp 130−133 °C;

Scheme 5. Introduction of a C Unit after Reductive C−N Bond Cleavage

 $[\alpha]^{27}$ _D = +1.3° (c 1.00, CHCl₃); IR (ATR) 1719 cm⁻¹; ¹H NMR (400 MHz) δ 1.40 (s, 9H), 2.69 (d, J = 2.2 Hz, 1H), 3.21 (d, J = 2.2 Hz, 1H), 3.82 (s, 3H), 3.84 (s, 6H), 4.08 (d, J = 14.0 Hz, 1H), 4.23 (d, J = 14.0 Hz, 1H), 6.53 (s, 2H), 7.24 (t, J = 6.0 Hz, 1H), 7.31 (t, J = 7.4 Hz, 2H), 7.40 $(d, J = 7.2 \text{ Hz}, 2\text{H})$; ¹³C NMR (100 MHz) δ 28.0, 45.7, 48.1, 54.6, 56.1, 60.8, 81.8, 103.0, 126.9, 128.1, 128.2, 134.1, 137.3, 139.2, 153.3, 167.6; HPLC (CHIRALCEL OD-H, λ 254 nm, hexane/ⁱPrOH 100/1, flow rate 1.0 mL/min) t_R for a major isomer 12.1 min, t_R for a minor isomer 17.9 min. Anal. Calcd for C₂₃H₂₉NO₅: C, 69.15; H, 7.32; N, 3.51. Found: C, 69.37; H, 7.38; N, 3.51.

cis-(2S,3S)-tert-Butyl 3-(3,4,5-trimethoxyphenyl)aziridine-2-carboxylate (cis-(2S,3S)-6): colorless oil; $[\alpha]^{27}$ _D = +3.3° (c 1.09, CHCl₃); IR (ATR) 1731 cm⁻¹; ¹H NMR (400 MHz) δ 1.21 (s, 9H), 2.53 (d, J = 7.0 Hz, 1H), 2.93 (d, J = 7.0 Hz, 1H), 3.52 (d, J = 13.8 Hz, 1H), 3.80 (s, 3H), 3.83 (s, 6H), 4.03 (d, J = 13.8 Hz, 1H), 6.63 (s, 2H), 7.26–7.46 (m, 5H); 13C NMR (100 MHz) δ 27.8, 46.8, 47.1, 56.0, 60.7, 63.1, 81.0, 104.6, 127.1, 127.9, 128.3, 131.1, 137.0, 137.9, 152.7, 167.0; HRMS (FAB) calcd for $C_{23}H_{30}NO_5$ 400.2124, found 400.2123; HPLC (CHIRALCEL OD-H, λ 254 nm, hexane/PrOH 50/1, flow rate 1.0 mL/min) t_R for a major isomer 15.6 min, t_R for a minor isomer 13.5 min.

(±)-tert-Butyl 2-Benzylamino-3-(2-benzyloxy-4,5-methylenedioxyphenyl)-3-(3,4,5-trimethoxyphenyl)propanoate $((\pm)$ -11a). A solution of (\pm) -trans-6 (104 mg, 0.259 mmol), InCl₃ (27 mg, 0.122 mmol), and sesamol benzyl ether (68 mg, 0.297 mmol) in CH₂Cl₂ (1.2 mL) was stirred at 35 $^{\circ}$ C for 2 h, quenched by addition of saturated NaHCO₃ solution (2 mL), and extracted with AcOEt (10 mL \times 3). The combined organic solutions were washed with H_2O (1 mL \times 2) and brine $(1 \text{ mL} \times 2)$, dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt 4/1) gave (\pm) -11a (89 mg, 55%, dr 5.6/1),

which was recrystallized from hexane/Et₂O ($1/1$) to give a single isomer (60 mg, 37%) as colorless prisms, mp 125−126 °C: IR (ATR) 3676, 1721 cm⁻¹; ¹H NMR (400 MHz) δ 1.18 (s, 9H), 3.61 (d, J = 13.4 Hz, 1H), 3.63 (s, 6H) 3.75 (s, 3H), 3.77 (d, J = 10.5 Hz, 1H), 3.85 (d, J = 13.4 Hz, 1H), 4.48 (d, J = 10.5 Hz, 1H), 4.85 (d, J = 11.5 Hz, 1H), 4.94 $(d, J = 11.5 Hz, 1H), 5.89 (d, J = 1.5 Hz, 1H), 5.94 (d, J = 1.5 Hz, 1H),$ 6.38 (s, 2H), 6.54 (s, 1H), 6.73 (s, 1H), 7.23−7.33 (m, 10H); 13C NMR (100 MHz) δ 27.7, 47.6, 51.9, 55.8, 60.7, 64.1, 71.1, 80.8, 96.0, 101.0, 106.1, 107.6, 122.4, 127.0, 127.4, 127.8, 128.2, 128.4, 128.5, 136.4, 136.8, 137.0, 139.6, 141.3, 146.3, 151.4, 152.6, 173.8; HRMS (FAB) calcd for $C_{37}H_{42}NO_8$ 628.2911, found 628.2889. Anal. Calcd for $C_{37}H_{42}NO_8$: C, 70.79; H, 6.58; N, 2.23. Found: C, 70.67; H, 6.60; N, 2.26.

(±)-tert-Butyl 2-Benzylamino-3-(2-allyloxy-4,5-methylenedioxyphenyl)-3-(3,4,5-trimethoxyphenyl)propanoate $((\pm)$ -11b). A solution of (\pm) -trans-6 (101 mg, 0.252 mmol), InCl₃ (18) mg, 0.08 mmol), and sesamol allyl ether (195 mg, 1.10 mmol) in CH_2Cl_2 (1.0 mL) was stirred at 30 °C for 3.5 h, quenched by addition of saturated NaHCO₃ solution (2 mL), and extracted with AcOEt (15 mL \times 3). The combined organic solutions were washed with $H_2O(1 \text{ mL} \times 2)$ and brine $(1 \text{ mL} \times 2)$, dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt 4/1) gave (\pm)-11b (87 mg, 60%, dr 4.3/1), which was recrystallized from hexane/Et₂O $(1/1)$ to give a single isomer (46 mg, 32%) as colorless prisms, mp 113−114 °C: IR (ATR) 3676, 1721 cm⁻¹; ¹H NMR (400 MHz) δ 1.20 (s, 9H), 3.61 (d, J = 13.4 Hz, 1H), 3.75−3.81 (m, 10H), 3.84 (d, J = 13.4 Hz, 1H), 4.32−4.46 (m, 2H), 4.52 $(d, J = 10.6 \text{ Hz}, 1H), 5.19 - 5.40 \text{ (m, 2H)}, 5.88 \text{ (d, J = 1.5 Hz, 1H)}, 5.92 \text{ (d,$ J = 1.5 Hz, 1H), 5.92−6.03 (m, 1H), 6.47 (s, 1H), 6.49 (s, 2H), 6.71 (s, 1H), 7.16−7.33 (m, 5H); 13C NMR (100 MHz,) δ 27.8, 51.9, 56.1, 60.7, 64.4, 65.8, 70.1, 80.9, 96.2, 101.5, 106.2, 107.7, 117.2, 122.7, 127.0, 128.2,

Scheme 6. Chemical Manipulation Trials for Preparation of 4-Aryltetralin Systems

Scheme 7. Preparation of Meyers' Intermediate 2

128.4, 133.5, 136.7, 139.8, 141.3, 146.4, 146.4, 151.4, 152.7, 173.9. Anal. Calcd for C₃₃H₃₉NO₈: C, 68.61; H, 6.80; N, 2.42. Found: C, 68.56; H, 6.80; N, 2.29.

(2S,3R)-tert-Butyl 2-Benzylamino-3-(2-hydroxy-4,5-methylenedioxyphenyl)-3-(3,4,5-trimethoxyphenyl)propanoate (11c). A solution of trans-(2S,3R)-6 (50 mg, 0.125 mmol), $\text{Zn}(\text{OTf})_2$ (4 mg,

Scheme 8. Preparation of ent-Zhang's Intermediate 3

0.011 mmol), and sesamol (21 mg, 0.153 mmol) in CH_2Cl_2 (1.2 mL) was stirred at 30 °C for 9 h, quenched by addition of saturated NaHCO₃ solution $(2 mL)$, and extracted with AcOEt $(15 mL \times 3)$. The combined organic solutions were washed with H₂O (5 mL \times 2) and brine (5 mL \times 2), dried, and evaporated. Column chromatography of the residue $(SiO₂)$ hexane/AcOEt $6/1$) gave 11c (60 mg, 89%, dr 10/1) as colorless prisms, mp 145−147 °C, which were twice washed with hexane/Et₂O (1/1): $[\alpha]^{27}$ _D = +17.1° (c 1.09, CHCl₃); IR (ATR) 3278, 1722 cm⁻¹; ¹H NMR (400 MHz) δ 1.29 $(s, 9H_1)$, 3.48 $(d, J = 13.4 \text{ Hz}, 1H)$, 3.54 $(d, J = 13.4 \text{ Hz})$ Hz, 1H), 3.80 (s, 6H), 3.82 (s, 3H), 3.84 (d, J = 5.6 Hz, 1H), 4.41 (d, J = 5.6 Hz, 1H), 5.85 (s, 2H), 6.42 (s, 1H), 6.52 (s, 1H), 6.56 (s, 2H), 7.09 $(d, J = 6.8 \text{ Hz}, 2\text{H}), 7.25-7.55 \text{ (m, 3H)}; {}^{13}\text{C} \text{ NMR} \text{ (100 MHz)} \overset{\circ}{\delta} 27.7,$ 52.2, 54.4, 56.0, 60.9, 64.8, 82.6, 100.3, 100.9, 105.5, 111.2, 115.7, 127.7, 128.1, 128.7, 134.8, 136.8, 137.4, 140.2, 147.8, 152.4, 153.3, 170.1; HRMS (FAB) calcd for $C_{30}H_{36}NO_8$ 538.2441, found 538.2427.

Ring Opening of the Methylenedioxyphenylaziridine (\pm)-trans-12 with 2,6-Dimethoxyphenol. A mixture of (\pm) -trans-12 (102 mg, 0.290 mmol), $InCl₃$ (6 mg, 0.029 mmol), and 2,6dimethoxyphenol (67 mg, 0.433 mmol) in CH_2Cl_2 (3 mL) was stirred at 40 °C for 14 h, quenched by addition of saturated NaHCO₃ solution (1 mL) , and extracted with AcOEt $(15 \text{ mL} \times 3)$. The combined organic solutions were washed with H₂O (1 mL \times 2) and brine (1 mL \times 2), dried, and evaporated. Column chromatography of the residue $(SiO₂)$ hexane/AcOEt 8/1) gave 2-aminopropanoate (\pm) -14a (32 mg, 22%) and the isomerized *cis*-aziridine (\pm) -*cis*-12 (32 mg, 31%).

(±)-tert-Butyl 2-benzylamino-3-[(2,6-dimethoxyphenyl)oxy]-3- (3,4-methylenedioxyphenyl)propanoate ((±)-14a): colorless oil; IR (ATR) 1729 cm⁻¹; ¹H NMR (400 MHz) δ 1.20 (s, 9H), 3.658 (s, 6H), 3.660 (d, J = 6.7 Hz, 1H), 3.78 (d, J = 13.0 Hz, 1H), 3.94 (d, J = 13.0 Hz, 1H), 5.30 (d, J = 6.7 Hz, 1H), 5.90 (s, 2H), 6.44 (d, J = 8.4 Hz, 2H), 6.66 $(d, J = 8.1 \text{ Hz}, 1\text{H})$, 6.83 $(dd, J = 8.1, 1.6 \text{ Hz}, 1\text{H})$, 6.88 $(t, J = 8.4 \text{ Hz},$ 1H), 7.09 (d, J = 1.6 Hz, 1H), 7.18−7.36 (m, 5H); 13C NMR (100 MHz) δ 27.7, 52.4, 55.7, 67.9, 80.8, 84.5, 100.7, 105.2, 107.1, 109.1, 122.0, 123.5, 126.7, 128.2, 128.3, 132.9, 135.8, 140.3, 146.8, 147.1, 153.2, 171.2; HRMS (ESI) calcd for $C_{29}H_{34}NO_7$ 508.2335, found 508.2318.

(±)-cis-tert-Butyl 3-(3,4-methylenedioxyphenyl)aziridine-2-carboxylate ((±)-cis-12): mp 176—178 °C; IR (ATR) 1731 cm⁻¹; ¹H NMR (400 MHz) δ 1.24 (s, 9H), 2.48 (d, J = 7.0 Hz, 1H), 2.93 (d, J = 6.8 Hz, 1H), 3.53 (d, J = 14.0 Hz, 1H), 3.96 (d, J = 14.0 Hz, 1H), 5.90 (s, 2H), 6.71 (d, J = 8.0 Hz, 1H), 6.86 (d, J = 8.0 Hz, 1H), 6.92 (s, 1H), 7.22−7.44 (m, 5H); 13C NMR (100 MHz) δ 27.8, 46.8, 47.1, 63.3, 81.1, 100.8, 107.7, 108.6, 121.2, 127.1, 127.8, 128.3, 129.3, 138.0, 146.7, 147.0, 167.2; HRMS (ESI) calcd for $C_{21}H_{23}NaNO_4$ 376.1525, found 376.1522.

(±)-tert-Butyl 2-Benzylamino-3-[(2,6-dimethoxyphenyl) amino]-3-(3,4-methylenedioxyphenyl)propanoate $((\pm)$ -14b). A mixture of (\pm) -trans-12 (64 mg, 0.181 mmol), InCl₃ (2 mg, 0.01 mmol), and 2,6-dimethoxyaniline (27 mg, 0.178 mmol) in CH_2Cl_2 (1.0 mL) was stirred at room temperature for 1.5 h, quenched by addition of saturated NaHCO₃ solution (3 mL), and extracted with AcOEt (5 mL \times 3). The combined organic solutions were washed with H₂O (2 mL \times 2) and brine $(2 mL \times 2)$, dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt 4/1) gave (\pm)-14b (83 mg, 92%) as colorless prisms, mp 97–99 °C: IR (ATR) 3334, 1711 cm^{−1}; ¹H NMR (400 MHz) δ 1.41 (s, 9H), 3.63 (d, J = 4.0 Hz, 1H), 3.66 (d, J = 13.1 Hz, 1H), 3.82 (s, 6H), 3.91 (d, J = 13.1 Hz, 1H), 5.25 (br s, 1H), 5.71 (br, 1H, exchangeable), 5.82 (s, 2H), 6.45 (d, J = 8.2 Hz, 2H), 6.58 (d, J = 8.1 Hz,

1H), 6.63 (dd, $I = 8.1$, 1.5 Hz, 1H), 6.69 (t, $I = 8.2$ Hz, 1H), 6.73 (d, $J = 1.5$ Hz, 1H), 7.24–7.27 (m, 1H), 7.34 (t, $J = 7.5$ Hz, 2H), 7.44 (d, $J =$ 7.5 Hz, 2H); ¹³C NMR (100 MHz) δ 28.1, 53.1, 55.8, 60.0, 66.1, 81.4, 100.6, 104.7, 107.5, 107.9, 119.3, 121.0, 125.8, 127.0, 128.27, 128.29, 133.8, 140.4, 146.4, 147.1, 150.6, 171.9; HRMS (ESI) calcd for $C_{29}H_{34}N_2NaO_6$ 529.2315, found 529.2334.

tert-Butyl 2-Benzylamino-3-[4,5-methylenedioxy-2-(trifluoromethanesulfonyloxy)phenyl]-3-(3,4,5-trimethoxyphenyl) **propanoate (15).** Tf₂O (0.2 mL, 1.17 mmol) was added dropwise to a stirred solution of 11c (416 mg, 0.773 mmol) and pyridine (0.1 mL, 1.24 mmol) in $\mathrm{CH_2Cl_2}$ (4.0 mL) at 0 $^\circ\mathrm{C}$, and the mixture was stirred at 0 °C for 5 h. After addition of saturated NaHCO₃ solution (10 mL), the mixture was extracted with AcOEt (30 mL \times 3). The combined organic solutions were washed with H₂O (5 mL \times 2) and brine (5 mL \times 2), dried, and evaporated. Column chromatography of the residue $(SiO₂)$, hexane/AcOEt 8/1) gave 15 (436 mg, 84%) as colorless prisms, mp 133–134 °C: IR (ATR) 1726 cm⁻¹; ¹H NMR (400 MHz) δ 1.26 (s, 9H), 3.62 (d, J = 13.5 Hz, 1H), 3.64 (d, J = 9.9 Hz, 1H), 3.78 (s, 3H), 3.80 (s, 6H), 3.85 (d, J = 13.5 Hz, 1H), 4.39 (d, J = 9.9 Hz, 1H), 6.00 (d, $J = 1.3$ Hz, 1H), 6.05 (d, $J = 1.3$ Hz, 1H), 6.52 (s, 2H), 6.73 (s, 1H), 6.78 (s, 1H), 7.16−7.32 (m, 5H); 13C NMR (100 MHz) δ 27.7, 47.1, 51.6, 56.1, 60.7, 64.5), 81.5, 102.4, 102.6, 105.9, 108.0, 116.9, 120.0, 127.1, 128.2, 135.0, 137.1, 139.3, 141.3, 146.8, 147.3, 153.0, 172.7; HRMS (ESI) calcd for $C_{31}H_{34}F_3NNaO_{10}S$ 692.1753, found 692.1733.

tert-Butyl 2-Benzylamino-3-(4,5-methylenedioxy-2-vinylphenyl)-3-(3,4,5-trimethoxyphenyl)propanoate (16). Tributylvinyltin (0.1 mL, 0.342 mmol) was added dropwise to a stirred mixture of 15 (200 mg, 0.299 mmol), $Pd(PPh_3)_4$ (35 mg, 0.030 mmol), and LiCl (25 mg, 0.599 mmol) in DMF (3.0 mL) at room temperature, and the mixture was stirred at 90 °C for 8 h. After addition of 10% KF solution (2 mL), the mixture was stirred for 30 min and filtered through Celite, which was washed with AcOEt (10 mL). After separation of the organic solution the aqueous solution was extracted with AcOEt (20 mL \times 2). The combined organic solutions were washed with $H_2O(2mL \times 2)$ and brine $(2 mL \times 2)$, dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt $6/1$) gave 16 (105 mg, 64%) as colorless prisms, mp 145−147 °C: IR (ATR) 3671, 1716 cm⁻¹; ¹H NMR (400 MHz) δ 1.22 (s, 9H), 3.58 (d, J = 13.0 Hz, 1H), 3.73 (d, J = 9.9 Hz, 1H), 3.77 (s, 3H), 3.78 (s, 6H), 3.81 (d, $J = 13.0$ Hz, 1H), 4.39 $(d, J = 9.9 \text{ Hz}, 1\text{ H}), 5.22 (dd, J = 10.8, 1.3 \text{ Hz}, 1\text{ H}), 5.45 (dd, J = 17.2, 1.3$ Hz, 1H), 5.92 (d, $J = 1.5$ Hz, 1H), 5.97 (d, $J = 1.5$ Hz, 1H), 6.43 (s, 2H), 6.80 (s, 1H), 6.92 (s, 1H), 7.05 (dd, J = 17.2, 10.8 Hz, 1H), 7.18–7.33 $(m, 5H);$ ¹³C NMR (100 MHz) δ 27.8, 49.3, 52.0, 56.1, 60.8, 65.1, 81.2, 101.0, 106.0, 106.5, 107.2, 115.4, 127.0, 128.2, 128.4, 131.6, 132.3, 134.6, 136.5, 139.5, 146.4, 147.4, 152.9, 173.5; HRMS (ESI) calcd for $C_{32}H_{38}NO_7$ 548.2648, found 548.2666.

tert-Butyl 2-(tert-Butoxycarbonylamino)-3-[4,5-methylenedioxy-2-(trifluoromethanesulfonyloxy)phenyl]-3-(3,4,5 trimethoxyphenyl)propanoate (17). A mixture of 15 (202 mg, 0.301 mmol), Pd/C (21 mg), and Boc₂O (0.15 mL, 0.653 mmol) in AcOEt (2.0 mL) was vigorously stirred at room temperature for 6 h under a hydrogen atmosphere. The mixture was filtered through Celite, which was washed with AcOEt. After evaporation of the filtrate, column chromatography of the residue (SiO₂, hexane/AcOEt 4/1) gave 17 (204 mg, 100%) as colorless prisms, mp 65−67 °C: IR (ATR) 3649, 1734, 1716 cm[−]¹ ; 1 H NMR (400 MHz) δ 1.22 (s, 9H), 1.35 (s, 9H), 3.80 (s,

3H), 3.83 (s, 6H), 4.44 (d, J = 9.9 Hz, 1H), 4.90 (t, J = 9.9 Hz, 1H), 4.97 (d, J = 9.5 Hz, 1H), 6.01 (s, 2H), 6.56 (s, 2H), 6.72 (s, 1H), 7.03 (s, 1H); 13 C NMR (100 MHz) δ 27.5, 28.1, 47.6, 56.0, 56.8, 60.8, 80.1, 82.1, 102.5, 102.8, 105.6, 108.1, 115.3, 117.4, 119.5, 121.6, 127.0, 133.8, 137.3, 140.9, 147.3, 147.6, 153.1, 155.1, 170.3; HRMS (FAB) calcd for $C_{29}H_{36}F_3NNaO_{12}S$ 702.1808, found 702.1779.

Reductive C−N Bond Cleavage of Aminopropanoate 11c. A 0.1 M solution of $SmI₂$ in THF (125 mL, 12.3 mmol) was prepared with CH_2I_2 (1.0 mL, 12.3 mmol), Sm (2.132 g, 14.2 mmol), and THF (125 mL) at room temperature. To the solution were successively added HMPA (2.2 mL, 12.4 mmol), H₂O (0.3 mL, 16.7 mmol), and a solution of 11c (1.36 g, 2.52 mmol) in THF (7 mL), and the mixture was stirred at room temperature for 1 h and then for 10 min in air after addition of H2O (20 mL). The precipitates were removed by filtration through Celite and repeatedly washed with AcOEt. After evaporation of the combined filtrates, the residue was diluted with AcOEt (200 mL). The organic solution was washed with 10% HCl solution $(20 \text{ mL} \times 3)$, H₂O (10 mL \times 2), and brine (10 mL \times 2), dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt $3/1$) gave 18 (1.01 g, 93%). The aqueous layer was made alkaline (pH >12) by addition of 10% NaOH solution and extracted with AcOEt (50 mL \times 2). The combined organic solutions were washed with H_2O (10 mL \times 2) and brine (10 mL \times 2), dried (K₂CO₃), and evaporated. Column chromatography of the residue (NH-SiO₂, hexane/CHCl₃ 1/1) gave $BnNH₂$ (250 mg, 93%) as a colorless oil.

(R)-tert-Butyl 3-(2-hydroxy-4,5-methylenedioxyphenyl)-3-(3,4,5-trimethoxyphenyl)propanoate (18): colorless needles, mp 175−177 °C; $[\alpha]^{27}$ _D = -22.2° (c 1.01, CHCl₃); IR (ATR) 3422, 1716 cm⁻¹; ¹H NMR (400 MHz) δ 1.39 (s, 9H), 2.92 (dd, J = 16.4, 10.4 Hz, 1H), 3.02 (dd, J = 16.4, 4.8 Hz, 1H), 3.82 (s, 3H), 3.83 (s, 6H), 4.66 (dd, J = 10.4, 4.8 Hz, 1H), 5.85 (d, J = 1.4 Hz, 1H), 5.86 (d, J = 1.4 Hz, 1H), 6.43 (s, 2H), 6.45 $(s, 1H)$, 6.48 $(s, 1H)$, 6.70 $(s, 1H)$; ¹³C NMR (100 MHz) δ 27.9, 39.1, 41.4, 56.1, 60.8, 82.0, 100.0, 101.0, 104.8, 107.2, 123.2, 136.6, 139.0, 141.8, 146.4, 148.0, 153.2, 173.4. Anal. Calcd for $C_{23}H_{28}O_8$: C, 63.88; H, 6.53. Found: C, 63.50; H, 6.56.

(R)-tert-Butyl 3-[4,5-Methylenedioxy-2-(trifluoromethanesulfonyloxy)phenyl]-3-(3,4,5-trimethoxyphenyl)propanoate (19). Tf₂O (0.03 mL, 0.175 mmol) was added dropwise to a stirred solution of 18 (50 mg, 0.116 mmol) and pyridine (0.03 mL, 0.372 mmol) in CH_2Cl_2 (1.2 mL) at −40 °C, and the mixture was stirred at 0 °C for 1 h. After addition of saturated NaHCO₃ solution (1 mL), the mixture was extracted with AcOEt (15 mL \times 2). The combined organic solutions were washed with H₂O (1 mL) and brine (1 mL \times 2), dried, and evaporated. Column chromatography of the residue $(SiO₂)$ hexane/ AcOEt 6/1) gave 19 (60 mg, 92%) as a colorless oil: $[\alpha]_{\text{D}}^{25} = -10.4^{\circ}$ $(c \ 1.02, \ \, \text{CHCl}_3)$; IR (ATR) 1718 cm⁻¹; ¹H NMR (400 MHz) δ 1.32 $(s, 9H)$, 2.81 (dd, J = 15.2, 7.8 Hz, 1H), 2.89 (dd, J = 15.2, 8.8 Hz, 1H), 3.81 (s, 3H), 3.83 (s, 6H), 4.72 (t, J = 8.2 Hz, 1H), 6.01 (d, J = 1.2 Hz, 1H), 6.02 (d, J = 1.2 Hz, 1H), 6.48 (s, 2H), 6.74 (s, 1H), 6.75 (s, 1H); ¹³C NMR (100 MHz) δ 27.6, 40.4, 41.4, 56.0, 60.7, 80.8, 102.4, 102.8, 104.7, 107.5, 116.8, 120.0, 129.9, 136.8, 136.9, 140.1, 146.8, 147.5, 153.1, 169.8; HRMS (ESI) calcd for $C_{24}H_{27}F_3NaO_{10}S$ 587.1175, found 587.1151.

(R)-tert-Butyl 3-(4,5-Methylenedioxy-2-vinylphenyl)-3-(3,4,5 trimethoxyphenyl)propanoate (4). Tributylvinyltin (0.5 mL, 1.71 mmol) was added dropwise to a stirred solution of 19 (769 mg, 1.36 mmol), $Pd(dppf)Cl_2\text{-}CH_2Cl_2$ (110 mg, 0.135 mmol), and LiCl (118 mg, 2.79 mmol) in DMF (7.7 mL) at room temperature, and the mixture was stirred at 50 °C for 7 h. After addition of 10% KF solution (5 mL), the mixture was stirred at room temperature for 30 min and filtered through Celite, which was washed with AcOEt. After separation of the organic solution the aqueous solution was extracted with AcOEt (50 mL \times 2). The combined organic solutions were washed with H_2O (5 mL \times 2) and brine (5 mL \times 2), dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt 6/1) gave 4 (541 mg, 90%, 81% ee) as colorless needles, mp 115−116 °C, which were twice recrystallized from hexane/MeOH (10) to give optically pure 4 $(338$ mg, 56%, 99% ee) as colorless needles, mp 115−116 °C: [α]²⁵_D = −20.2° (c 1.02, CHCl₃); IR (ATR) 1717 cm[−]¹ ; 1 H NMR (400 MHz) δ 1.31 (s, 9H), 2.81 (dd, J = 15.3, 7.7 Hz, 1H), 2.88 (dd, J = 15.3, 8.4 Hz, 1H), 3.80 (s, 9H), 4.72

 $(t, J = 8.0 \text{ Hz}, 1\text{ H}), 5.26 \text{ (dd, } J = 11.2, 1.3 \text{ Hz}, 1\text{ H}), 5.51 \text{ (dd, } J = 17.2, 1.3$ Hz, 1H), 5.92 (d, $J = 1.5$ Hz, 1H), 5.93 (d, $J = 1.5$ Hz, 1H), 6.41 (s, 2H), 6.66 (s, 1H), 6.96 (s, 1H), 7.09 (dd, J = 17.2, 11.2 Hz, 1H); ¹³C NMR (100 MHz) δ 27.9, 42.1, 42.5, 56.1, 60.8, 80.7, 101.0, 104.8, 106.3, 107.2, 115.2, 130.4, 134.3, 134.7, 136.5, 138.8, 146.3, 147.5, 153.1, 170.9; HRMS (ESI) calcd for $C_{25}H_{30}NaO_7$ 465.1889, found 465.1878; HPLC (CHIRALCEL OD-H, λ 254 nm, hexane/ⁱ PrOH 9/1, flow rate 1.0 mL/ min) t_R for the major isomer 9.3 min, t_R for the minor isomer 10.9 min.

trans-Methyl 2-[2-(tert-Butoxycarbonyl)-1-(3,4,5 trimethoxyphenyl)ethyl]-4,5-methylenedioxyphenylpropenoate (20). Methyl acrylate (0.05 mL, 0.555 mmol) was added dropwise to a stirred mixture of 19 (50 mg, 0.0893 mmol), $Pd(PPh₃)₄$ $(17 \text{ mg}, 0.015 \text{ mmol})$, and NaHCO₃ $(39 \text{ mg}, 0.458 \text{ mmol})$ in DMF (1.0 mL) at room temperature, and the mixture was stirred at 120 °C for 24 h. After addition of $H₂O$ (2 mL), the mixture was extracted with AcOEt (15 mL \times 3). The combined organic solutions were washed with $H₂O$ (2 mL \times 3) and brine (2 mL \times 3), dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt $3/1$) gave 20 (27 mg, 61%) as colorless prisms, mp 168−170 °C: IR (ATR) 1717 cm⁻¹;
¹H NMR (400 MHz) δ 1 31 (s 9H) 2 85 (dd I = 15 2 8 1 Hz 1H) 2 92 1 H NMR (400 MHz) δ 1.31 (s, 9H), 2.85 (dd, J = 15.2, 8.1 Hz, 1H), 2.92 $(dd, I = 15.2, 8.1 Hz, 1H), 3.78 (s, 3H), 3.80 (s, 3H), 3.81 (s, 6H), 4.79$ $(t, J = 8.1 \text{ Hz}, 1\text{ H}), 5.98 \text{ (d, } J = 1.3 \text{ Hz}, 1\text{ H}), 5.99 \text{ (d, } J = 1.3 \text{ Hz}, 1\text{ H}), 6.18$ $(d, J = 15.7 \text{ Hz}, 1\text{H})$, 6.41 (s, 1H), 6.79 (s, 1H), 7.01 (s, 1H), 8.17 (d, J = 15.7 Hz, 1H); 13C NMR (100 MHz) δ 27.9, 42.2, 42.8, 51.6, 56.1, 60.8, 80.9, 101.5, 104.7, 106.4, 107.2, 118.0, 126.8, 136.5, 137.8, 138.4, 141.8, 146.6, 149.6, 153.2, 167.3, 170.4; HRMS (FAB) calcd for $C_{27}H_{32}O_9$ 500.2046, found 500.2043.

tert-Butyl 3-[4,5-Methylenedioxy-2-(oxacyclopropyl) phenyl]-3-(3,4,5-trimethoxyphenyl)propanoate (21). A solution of 4 (100 mg, 0.226 mmol) and NIS (61 mg, 0.270 mmol) in THF/H₂O (4/1, 1.0 mL) was stirred at room temperature for 4 h. After addition of $K₂CO₃$ (81 mg, 0.589 mmol) and MeOH (1.0 mL), the mixture was stirred at room temperature for 1 h, quenched by addition of saturated $Na₂S₂O₃$ solution (1 mL), and extracted with AcOEt (20 mL \times 2). The combined organic solutions were washed with H₂O (1 mL \times 2), dried, and evaporated to give 21 (97 mg, 93%, dr $2/1$) as pale yellow needles, mp 78−82 °C, which were used in the next reaction without further purification: IR (ATR) 1717 cm⁻¹; ¹H NMR (400 MHz) for the major isomer δ 1.33 (s, 9H), 2.39 (dd, J = 5.7, 2.7 Hz, 1H), 2.92 (d, J = 8.1 Hz, 2H), 2.95 (dd, J = 5.7, 4.1 Hz, 1H), 3.79 (s, 6H), 3.803 (s, 3H), 4.12 (dd, $J = 4.1, 2.7$ Hz, 1H), 4.74 (t, $J = 8.1$ Hz, 1H), $5.92 - 5.95$ (m, 2H), 6.38 (s, 2H), 6.758 (s, 1H), 6.79 (s, 1H); ¹H NMR (400 MHz) for the minor isomer δ 1.31 (s, 9H), 2.70 (dd, J = 5.8, 2.7 Hz, 1H), 2.88 (dd, J = 8.0, 7.0 Hz, 2H), 3.17 (dd, J = 5.8, 4.2 Hz, 1H), 3.801 (s, 6H), 3.81 (s, 3H), 4.01 $(dd, J = 4.2, 2.7 Hz, 1H), 4.75 (t, J = 8.0 Hz, 1H), 5.92-5.95 (m, 2H),$ 6.46 (s, 2H), 6.75 (s, 1H), 6.762 (s, 1H); ¹³C NMR (100 MHz) for the major isomer δ 27.9, 42.2, 42.6, 50.3, 50.4, 56.1, 60.8, 80.8, 101.1, 104.8, 105.4, 106.9, 129.34, 135.1, 136.6, 138.9, 146.4, 147.25, 153.1, 170.78; ¹³C NMR (100 MHz) for the minor isomer δ 27.9, 42.3, 42.5, 50.3, 50.4, 56.1, 60.7, 80.7, 101.1, 104.9, 105.4, 107.0, 129.28, 135.2, 136.7, 138.5, 146.3, 147.27, 153.2, 170.81; HRMS (ESI) calcd for $C_{25}H_{30}NaO_8$ 481.1838, found 481.1840.

(R)-tert-Butyl 3-{2-[(1-Hydroxy-2-iodo)ethyl]-4,5-methylenedioxyphenyl}-3-(3,4,5-trimethoxyphenyl)propanoate (22). A solution of 4 (200 mg, 0.452 mmol) and NIS (121 mg, 0.537 mmol) in THF/H₂O (4/1, 2.0 mL) was stirred at room temperature for 7 h in air, quenched by addition of saturated $\text{Na}_2\text{S}_2\text{O}_3$ solution (1 mL), and extracted with AcOEt (20 mL \times 2). The combined organic solutions were washed with saturated $\text{Na}_2\text{S}_2\text{O}_3$ solution (1 mL), H₂O (1 mL \times 2), and brine $(1 mL \times 2)$, dried, and evaporated. Column chromatography of the residue $(SiO₂$, hexane/AcOEt 3/1) gave 22 (253 mg, 95%, dr 2/1) as colorless prisms, mp 182–184 °C: $[\alpha]^{25.5}$ _D = +22.1° (c 1.08, CHCl₃); IR (ATR) 3508, 1723 cm⁻¹; ¹H NMR (400 MHz) for the major isomer δ 1.32 (s, 9H), 2.71 (dd, J = 10.3, 3.2 Hz, 1H), 2.90 (d, J = 8.0 Hz, 2H), 3.00 (d, J = 2.4 Hz, 1H, exchangeable), 3.13 (t, J = 9.8 Hz, 1H), 3.796 $(s, 3H)$, 3.83 $(s, 6H)$, 4.61 $(t, J = 8.0$ Hz, 1H), 5.30 $(dt, J = 9.3, 3.2$ Hz, 1H), 5.96 (s, 2H), 6.40 (s, 2H), 6.81 (s, 1H), 7.00 (s, 1H); ¹ H NMR (400 MHz) for the minor isomer δ 1.34 (s, 9H), 2.57 (d, J = 3.1 Hz, 1H, exchangeable), 2.89 (d, J = 8.0 Hz, 2H), 3.30 (t, J = 10.0 Hz, 1H), 3.55 (dd, J = 10.3, 3.0 Hz, 1H), 3.803 (s, 3H), 3.81 (s, 6H), 4.54 (t, J = 8.0 Hz, 1H), 5.11 (dt, J = 9.6, 2.8 Hz, 1H), 5.96 (s, 2H), 6.37 (s, 2H), 6.83 (s, 1H), 7.03 (s, 1H); 13C NMR (100 MHz) for the major isomer δ 13.1, 27.9, 42.2, 42.8, 56.2, 60.8, 70.1, 81.1, 101.2, 104.6, 106.0, 106.6, 133.0, 133.5, 136.7, 139.0, 146.5, 147.5, 153.3, 171.1; 13C NMR (100 MHz) for the minor isomer δ 13.9, 27.9, 42.4, 42.5, 56.1, 60.7, 70.9, 81.0, 101.2, 104.8, 106.4, 107.2, 133.0, 133.2, 136.7, 138.6, 146.5, 147.4, 153.2, 170.8; HRMS (ESI) calcd for $C_{25}H_{31}IKO_8$ 625.0701, found 625.0709.

2-(tert-Butoxycarbonyl)-3-hydroxymethyl-5,6-methylenedioxy-1-(3,4,5-trimethoxyphenyl)indane (24). The reaction flask was dried by heat gun under reduced pressure for 1 h. A 1.6 M solution of LiHMDS in THF (0.16 mL, 0.256 mmol) was added dropwise to a stirred solution of epoxide 21 (50 mg, 0.109 mmol) in THF (2.2 mL) at −78 °C, and the mixture was stirred at −78 °C for 18 h. After addition of saturated NH₄Cl solution (1 mL) at 0 $^{\circ}{\rm C},$ the mixture was extracted with AcOEt (20 mL \times 2). The combined organic solutions were washed with $H₂O$ (1 mL \times 2) and brine (1 mL \times 2), dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt 3/1) gave 24 (15) mg, 30%) as a colorless needles, mp 66−68 °C: IR (ATR) 3504, 1723 cm⁻¹; ¹H NMR (400 MHz) δ 1.45 (s, 9H), 3.10 (t, J = 9.2 Hz, 1H), 3.47−3.54 (m,1H), 3.81 (s, 6H), 3.85 (s, 3H), 3.96 (dd, J = 11.2, 6.2 Hz, 1H), 4.06 (dd, J = 11.2, 4.2 Hz, 1H), 4.48 (d, J = 9.2 Hz, 1H), 5.92 (d, J = 1.3 Hz, 1H), 5.95 (d, J = 1.3 Hz, 1H), 6.40 (s, 1H), 6.43 (s, 2H), 6.74 (s, 1H); 13C NMR (100 MHz) δ 28.1, 49.9, 53.6, 56.1, 59.2, 60.8, 65.3, 81.4, 101.2, 103.6, 105.3, 105.6, 133.9, 136.9, 138.2, 139.4, 147.5, 147.7, 153.2, 173.6; HRMS (ESI) calcd for $C_{25}H_{30}NaO_8$ 481.1838, found 481.1827.

tert-Butyl 3-(2-Iodoacetyl-4,5-methylenedioxyphenyl)-3- (3,4,5-trimethoxyphenyl)propanoate (25). A mixture of 22 (50 mg, 0.085 mmol), DMP (54 mg, 0.128 mmol), and NaHCO₃ (22 mg, 0.256 mmol) in CH_2Cl_2 (0.9 mL) was stirred at room temperature for 4 h. After addition of saturated NaHCO₃ solution (1 mL) with ice cooling, the mixture was extracted with AcOEt (20 mL \times 2). The combined organic solutions were washed with H₂O (1 mL \times 2) and brine $(1 \text{ mL} \times 2)$, dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt 4/1) gave 25 (43 mg, 85%) as labile colorless prisms, mp 116−118 °C: ¹H NMR¹³ (400 MHz) δ 1.31 (s, 9H), 2.85 (dd, J = 15.6, 8.8 Hz, 1H), 2.93 (dd, J = 15.6, 7.3 Hz, 1H), 3.80 $(s, 3H)$, 3.83 $(s, 6H)$ [, 4](#page-11-0).23 $(d, J = 10.8 \text{ Hz}, 1H)$, 4.30 $(d, J = 10.8 \text{ Hz}, 1H)$, 5.05 (t, $I = 8.1$ Hz, $1H$), 5.99 (d, $I = 1.3$ Hz, $1H$), 6.01 (d, $I = 1.3$ Hz, $1H$), 6.53 (s, 2H), 6.78 (s, 1H), 7.04 (s, 1H).

(R)-tert-Butyl 3-{2-[1-(tert-Butyldimethylsilyloxy)-2-iodoethyl]- 4,5-methylenedioxyphenyl}-3-(3,4,5-trimethoxyphenyl) propanoate (27). TBSOTf (0.03 mL, 0.128 mmol) was added to a stirred solution of 22 (50 mg, 0.086 mmol, dr $2/1$) and 2,6-lutidine $(0.03 \text{ mL}, 0.259 \text{ mmol})$ in CH₂Cl₂ (0.9 mL) at 0 °C, and the mixture was stirred at room temperature for 8 h. After addition of saturated $NAHCO₃$ solution (1 mL) at $0 \text{ }^{\circ}\text{C}$, the mixture was extracted with AcOEt (20 mL \times 2). The combined organic solutions were washed with H₂O $(1 \text{ mL} \times 2)$ and brine $(1 \text{ mL} \times 2)$, dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt $6/1$) gave 27 (55 mg, 92%, dr 2/1) as a colorless oil: $[\alpha]_{D}^{26} = +104.1^{\circ}$ (c 0.954, CHCl₃); IR (ATR) 1731 cm⁻¹; ¹H NMR (400 MHz) for the major isomer δ −0.08 (s, 3H), 0.21 (s, 3H), 0.91 (s, 9H), 1.35 (s, 9H), 2.77− 2.93 (m, 4H), 3.783 (s, 3H), 3.83 (s, 6H), 4.52 (t, J = 7.8 Hz, 1H), 5.15 $(dd, J = 9.6, 1.7 Hz, 1H), 5.96 (d, J = 1.5 Hz, 1H), 5.99 (d, J = 1.5 Hz,$ 1H), 6.35 (s, 2H), 6.87 (s, 1H), 7.00 (s, 1H); ¹H NMR (400 MHz) for the minor isomer δ –0.24 (s, 3H), 0.10 (s, 3H), 0.79 (s, 9H), 1.35 (s, 9H), 2.34 (dd, J = 10.2, 1.9 Hz, 2H), 3.15 (dd, J = 10.4, 9.6 Hz, 1H), 3.57 $(dd, J=10.4, 2.0 Hz, 1H), 3.78 (s, 3H), 3.81 (s, 6H), 4.36 (dd, J=9.4, 6.5$ Hz, 1H), 4.96 (dd, J = 9.6, 2.0 Hz, 1H), 5.96 (d, J = 1.5 Hz, 1H), 6.01 (d, $J = 1.5$ Hz, 1H), 6.37 (s, 2H), 6.88 (s, 1H), 7.06 (s, 1H); ¹³C NMR (100 MHz) for the major isomer δ −4.77, −4.63, 14.3, 18.2, 25.8, 28.0, 43.1, 43.4, 56.2, 60.9, 71.0, 80.8, 101.1, 104.6, 106.3, 107.0, 132.0, 135.4, 136.7, 139.3, 146.4, 147.1, 153.4, 170.6; 13C NMR (100 MHz) for the minor isomer δ −5.43, −3.61, 14.1, 18.1, 25.7, 28.0, 43.0, 43.3, 56.1, 60.7, 67.9, 80.9, 101.1, 105.3, 106.2, 106.7, 131.4, 135.8, 136.8, 138.6, 146.3, 147.0, 153.4, 170.5; HRMS (ESI) calcd for $C_{31}H_{45}NaO_8Si$ 723.1826, found 723.1849.

(1R,2R)-2-(tert-Butoxylcarbonyl)-4-(tert-butyldimethylsilyloxy)-6,7-methylenedioxy-1-(3,4,5-trimethoxyphenyl)-1,2,3,4 tetrahydronaphthalene (28). The reaction flask was dried by heat gun under reduced pressure for 1 h. A 1.6 M solution of LHMDS in THF (0.1 mL, 0.16 mmol) was added dropwise to a stirred solution of 27 (65 mg, 0.092 mmol, dr 2/1) and HMPA (0.05 mL, 0.282 mmol) in THF (1.0 mL) at −78 °C, and the mixture was stirred at−78 °C for 10 h. After addition of saturated NH₄Cl solution (1 mL) at -78 °C, the mixture was extracted with AcOEt (20 mL \times 2). The combined organic solutions were washed with H₂O (1 mL \times 2) and brine (1 mL \times 2), dried, and evaporated. Column chromatography of the residue $(SiO₂)$, hexane/ AcOEt 6/1) gave 28 (47 mg, 90%, dr 2/1) as colorless needles, mp 69− 74 °C: $[\alpha]^{26.5}$ _D = +34.0° (c 1.04, CHCl₃); IR (ATR) 1727 cm⁻¹; ¹H NMR (400 MHz) for the major isomer δ 0.18 (s, 3H), 0.24 (s, 3H), 0.98 $(s, 9H)$, 1.27 $(s, 9H)$, 1.96–2.11 (m, 2H), 2.82 (ddd, J = 13.2, 11.2, 2.0 Hz, 1H), 3.78 (s, 3H), 3.795 (s, 6H), 4.09−4.15 (m, 1H), 4.94 (dd, J = 11.2, 4.8 Hz, 1H), 5.86−5.90 (m, 2H), 6.22 (s, 1H), 6.33 (s, 2H), 6.95 (s, 1H); ¹H NMR (400 MHz) for the minor isomer δ 0.15 (s, 3H), 0.18 (s, 3H), 0.92 (s, 9H), 1.32 (s, 9H), 2.25 (dd, J = 5.3, 2.6 Hz, 1H), 2.28 (dd, $J = 5.5, 2.4$ Hz, 1H), 3.18 (dt, $J = 10.7, 3.4$ Hz, 1H), 3.799 (s, 3H), 3.83 (s, 6H), 4.09−4.15 (m, 1H), 4.75 (t, J = 3.7 Hz, 1H), 5.86−5.90 (m, 2H), 6.32 (s, 1H), 6.37 (s, 2H), 6.67 (s, 1H); 13C NMR (100 MHz) for the major isomer δ −4.68, −3.92, 18.1, 25.9, 27.8, 36.7, 49.2, 49.5, 56.1, 60.9, 69.6, 80.5, 100.8, 106.0, 106.3, 108.8, 131.8, 133.6, 136.8, 139.5, 146.3, 146.6, 153.1, 173.1; ¹³C NMR (100 MHz) for the minor isomer δ –4.45, −4.26, 18.0, 25.8, 27.9, 35.1, 45.0, 47.9, 56.0, 60.8, 67.7, 80.2, 101.0, 106.2, 108.1, 109.3, 131.7, 132.0, 136.9, 139.5, 145.9, 147.2, 153.0, 174.5; HRMS (ESI) calcd for $C_{31}H_{44}NaO_8Si$ 595.2703, found: 595.2707.

(1R,2R)-2-(tert-Butoxylcarbonyl)-4-hydroxy-6,7-methylenedioxy-1-(3,4,5-trimethoxyphenyl)-1,2,3,4-tetrahydronaphthalene (23). A 1.0 M solution of TBAF in THF (0.67 mL, 0.67 mmol) was added to a solution of 28 (192 mg, 0.335 mmol, dr $2/1$) in THF (3 mL) at room temperature, and the mixture was stirred at the same temperature for 8 h in air. After addition of H_2O (3 mL), the mixture was extracted with AcOEt (30 mL \times 2). The combined organic solutions were washed with H₂O (2 mL \times 2) and brine (2 mL \times 2), dried, and evaporated. Column chromatography of the residue $(SiO₂)$, hexane/ AcOEt 3/1) gave 23 (139 mg, 90%, dr 2/1) as colorless needles, mp 65− 67 °C: $[\alpha]_{D}^{25}$ = +24.6° (c 0.97, CHCl₃); IR (ATR) 3733, 1717 cm⁻¹; ¹H NMR (400 MHz) for the major isomer δ 1.31 (s, 9H), 2.01 (ddd, J = 13.2, 10.0, 8.0 Hz, 1H), 2.35 (ddd, J = 13.2, 6.0, 3.2 Hz, 1H), 2.45 (d, J = 8.8 Hz, 1H), 2.85 (ddd, J = 10.0, 8.4, 3.2 Hz, 1H), 3.78 (s, 6H), 3.83 (s, 3H), 4.22 (d, J = 8.0 Hz, 1H), 4.80−4.86 (m, 1H), 5.90 (d, J = 1.2 Hz, 1H), 5.91 (d, J = 1.2 Hz, 1H), 6.26 (s, 2H), 6.39 (s, 1H), 7.08 (s, 1H); ¹H NMR (400 MHz) for the minor isomer δ 1.25 (s, 9H), 1.89 (d, J = 4.0 Hz, 1H), 2.13 (ddd, J = 13.6, 12.3, 3.2 Hz, 1H), 2.24 (dt, J = 13.6, 3.2 Hz, 1H), 3.13 (ddd, J = 12.3, 11.0, 2.8 Hz, 1H), 3.80 (s, 6H), 3.84 (s, 3H), 4.01 (d, J = 10.4 Hz, 1H), 4.80−4.86 (m, 1H), 5.90 (s,2H), 6.30 (s, 2H), 6.31 (s, 1H), 6.81 (s, 1H); ¹³C NMR (150 MHz) for the major isomer δ 27.9, 34.48, 48.0, 48.7, 56.15, 60.87, 67.8, 81.0, 101.0, 106.1, 106.9, 108.9, 130.9, 132.8, 136.80, 139.5, 146.7, 147.1, 153.1, 173.9; 13C NMR (150 MHz) for the minor isomer δ 27.8, 34.52, 44.6, 49.3, 56.17, 60.89, 67.1, 80.4, 101.1, 106.5, 108.5, 109.2, 130.4, 132.6, 136.85, 139.1, 146.4, 147.7, 153.1, 174.2; HRMS (ESI) calcd for $C_{25}H_{30}NaO_8$ 481.1838, found 481.1824.

(1R,2R)-2-(tert-Butoxylcarbonyl)-6,7-methylenedioxy-1- (3,4,5-trimethoxyphenyl)-2,3-dihydro-4(1H)-naphthalenone (26). A mixture of 23 (52 mg, 0.114 mmol, dr 2/1), DMP (66 mg, 0.155 mmol), and NaHCO₃ (25 mg, 0.299 mmol) in CH_2Cl_2 (1.1 mL) was stirred at room temperature for 3 h. After addition of saturated NaHCO_3 solution (1 mL) with ice cooling, the mixture was extracted with AcOEt (20 mL \times 2). The combined organic solutions were washed with H₂O $(1 \text{ mL} \times 2)$ and brine $(1 \text{ mL} \times 2)$, dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt 4/1) gave 26 (48 mg, 92%) as colorless prisms, mp 116–118 °C: $[\alpha]_{\text{D}}^{23} = +41.7^{\circ}$ $(c 1.01, CHCl₃)$; IR (ATR) 1721, 1666 cm⁻¹; ¹H NMR (400 MHz) δ 1.24 (s, 9H), 2.78 (dd, J = 17.0, 4.6 Hz, 1H), 2.87 (dd, J = 17.0, 9.0 Hz, 1H), 3.22 (ddd, J = 9.0, 8.0, 4.6 Hz, 1H), 3.79 (s, 6H), 3.84 (s, 3H), 4.40 $(d, J = 8.0 \text{ Hz}, 1\text{ H}), 6.00 \text{ (br s, 1H)}, 6.01 \text{ (br s, 1H)}, 6.34 \text{ (s, 2H)}, 6.41 \text{ (s, 2H)}$ 1H), 7.52 (s, 1H); 13C NMR (100 MHz) δ 27.7, 38.7, 48.6, 49.3, 56.2, 60.9, 81.4, 101.8, 105.8, 106.1, 108.8, 127.2, 136.9, 137.2, 140.8, 147.4, 152.5, 153.3, 171.8, 193.9; HRMS (ESI) calcd for $C_{2.5}H_{2.8}NaO_8$ 479.1682, found 479.1664.

Preparation of Meyers' Intermediate 2. A mixture of 26 (30 mg, 0.066 mmol), 37% HCHO solution (1.0 mL), and 1 N NaOH solution (0.14 mL, 0.14 mmol) in THF (1.0 mL) was stirred at room temperature for 1 day in air. After addition of saturated NH_4Cl solution (1 mL), the mixture was extracted with AcOEt (20 mL \times 2). The combined organic solutions were washed with H₂O (1 mL \times 2) and brine $(1 \text{ mL} \times 2)$, dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt 3/1) gave 2 (28 mg, 95%) as colorless prisms, mp 106−108 °C (lit.^{3g} mp 105−108 °C): $[\alpha]^{25}$ _D = −23.9° (c 0.98, CHCl₃) (lit.^{3g} $[\alpha]^{23}$ _D = −23.7° (c 0.38, CHCl₃)); IR (ATR) 3505, 2921, 1770, 1659, 1593, 1474 cm⁻¹; ¹H NMR (400 MHz) δ 3.26 (br d, J = 11.4 Hz, 1H), 3.36 (d, J = 2.2 Hz, 1H), 3.64 (d, J = 11.4 Hz, 1H), 3.74 (s, 6H), 3.81 (s, 3H), 4.35 (d, J = 9.4 Hz, 1H), 4.59 (d, J = 9.4 Hz, 1H), 4.78 (d, J = 2.2 Hz, 1H), 6.07 (s, 1H), 6.10 (s, 1H), 6.15 (s, 2H), 6.70 (s, 1H), 7.47 (s, 1H); 13C NMR (100 MHz) δ 42.8, 48.0, 54.0, 56.3, 60.9, 64.2, 72.7, 102.3, 105.1, 106.3, 109.7, 127.2, 137.3, 137.9, 138.8, 148.6, 153.6, 154.0, 176.0, 196.8; HRMS (ESI) calcd for $C_{23}H_{22}NaO_9$ 465.1162, found 465.1157.

tert-Butyl 2-Allyl-3-(4,5-methylenedioxy-2-vinylphenyl)-3- (3,4,5-trimethoxyphenyl)propanoate (29). The reaction flask was dried by heat gun under reduced pressure for 1 h. A 1.6 M solution of LHMDS in THF (0.15 mL, 0.24 mmol) was added dropwise to a stirred solution of 4 (50 mg, 0.113 mmol) in THF (2.2 mL) at -78 °C, and the mixture was stirred at −78 °C for 30 min. After addition of allyl bromide (0.03 mL, 0.344 mmol) at −78 °C, the mixture was stirred at −78 °C for 1 h and then at room temperature for 2.5 h. After addition of saturated $NH₄Cl$ solution $(2 mL)$ with ice cooling, the mixture was extracted with AcOEt (15 mL \times 3). The combined organic solutions were washed with $H₂O$ (1 mL \times 2) and brine (1 mL \times 2), dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt $6/1$) gave 29 (50 mg, 92%, dr 3/1) as a colorless oil: IR (ATR) 1724 cm $^{-1}$; ¹H NMR (400 MHz) for the major isomer δ 1.19 (s, 9H), 2.24 (t, J = 7.0 Hz, 1H), $3.05-3.15$ (m, 1H), 3.79 (s, 3H), 3.83 (s, 6H), 4.27 (d, J = 11.6 Hz, 1H), $4.96-5.05$ (m, 2H), 5.27 (dd, J = 10.9, 1.2 Hz, 1H), 5.44 (dd, J = 17.1, 1.2 Hz, 1H), $5.67 - 5.81$ (m, 1H), 5.88 (d, J = 1.3 Hz, 1H), 5.92 (d, J = 1.3 Hz, 1H), 6.47 (s, 2H), 6.87 (s, 1H), 6.97 (s, 1H), 7.19 (dd, J = 17.1, 10.9 Hz, 1H); ¹H NMR (400 MHz) for the minor isomer δ 1.21 $(s, 9H)$, 2.24 (t, J = 7.0 Hz, 1H), 3.05–3.15 (m, 1H), 3.76 (s, 3H), 3.83 (s, 6H), 4.39 (d, J = 11.7 Hz, 1H), 4.96–5.05 (m, 2H), 5.30 (dd, J = 10.7, 1.2 Hz, 1H), 5.44 (dd, J = 16.8, 1.2 Hz, 1H), 5.67−5.81 (m, 1H), 5.90 (d, $J = 1.2$ Hz, 1H), 5.95 (d, $J = 1.2$ Hz, 1H), 6.51 (s, 2H), 6.80 (s, 1H), 6.92 (s, 1H), 7.18–7.22 (m, 1H); ¹³C NMR (100 MHz) for the major isomer δ 27.7, 36.0, 48.5, 51.5, 56.1, 60.7, 80.5, 100.9, 105.0, 106.6, 106.9, 115.4, 116.8, 130.8, 134.1, 135.04, 135.1, 136.60, 137.4, 146.2, 147.3, 153.2, 172.7; 13C NMR (100 MHz) for the minor isomer δ 27.8, 35.6, 48.0, 51.4, 56.0, 60.7, 80.6, 101.1, 105.2, 106.2, 106.5, 115.6, 116.8, 131.3, 133.5, 134.8, 134.96, 136.56, 138.2, 146.3, 147.8, 152.9, 173.5; HRMS (ESI) calcd for C₂₈H₃₄NaO₇ 505.2202, found 505.2188.

Preparation of ent-Zhang's Intermediate 3. A solution of 29 $(31 \text{ mg}, 0.064 \text{ mmol}, \text{dr } 3/1)$, NaIO₄ $(82 \text{ mg}, 0.384 \text{ mmol})$, and 0.1 M OsO₄ solution (0.06 mL, 0.006 mmol) in 1,4-dioxane/H₂O (4/1, 2.0 mL) was stirred at 50 °C for 8 h in air, quenched by addition of saturated $\text{Na}_2\text{S}_2\text{O}_3$ solution (2 mL) with ice cooling, and extracted with AcOEt (15 mL \times 3). The combined organic solutions were washed with $H₂O$ (1 mL \times 2) and brine (1 mL \times 2), dried, and evaporated. Column chromatography of the residue (SiO₂, hexane/AcOEt $6/1$) gave 3 $(15 \text{ mg}, 49\%)$ as a colorless oil: $[\alpha]_{D}^{27} + 94.74^{\circ}$ (c 1.03, CHCl₃) (lit.^{3a}) $[\alpha]_{\text{D}}^{27}$ = -95.38° (c 1.66, CHCl₃)); IR (ATR) 1724, 1676 cm⁻¹; ¹H NMR (400 MHz) δ 1.19 (s, 9H), 2.60 (dd, J = 18.3, 3.7 Hz, 1H), 2.88 $(ddd, J = 18.3, 10.0, 1.3 Hz, 1H), 3.56–3.64 (m, 1H), 3.79 (s, 3H), 3.84$ $(s, 6H)$, 5.23 (d, J = 11.7 Hz, 1H), 6.02 (d, J = 1.6 Hz, 1H), 6.06 (d, J = 1.6 Hz, 1H), 6.52 (s, 2H), 7.14 (s, 1H), 7.24 (s, 1H), 9.67 (s, 1H), 10.3 (s, 1H); $^{13}\mathrm{C}$ NMR (100 MHz) δ 27.5, 44.9, 45.5, 45.6, 56.2, 60.7, 81.3, 102.1, 105.1, 108.3, 110.2, 128.2, 136.1, 137.2, 141.2, 146.9, 152.2, 153.5, 172.2, 189.6, 199.4; HRMS (ESI) calcd for $C_{26}H_{30}NaO_9$ 509.1788, found 509.1773.

■ ASSOCIATED CONTENT

S Supporting Information

Figures giving NMR spectra for compounds 2−4, trans-(2S,3R) and cis-(2S,3S)-6, (\pm) -11a,b, 11c, (\pm) -cis-12, (\pm) -14a,b, and 15−29 (only ¹ H NMR spectrum for 25) and HPLC traces for (\pm) -trans- and trans-(2S,3R)-6, (\pm) -cis- and cis-(2S,3S)-6, and (\pm) - and (R) -4. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The aut[hors declare no competin](mailto:benti@faculty.chiba-u.jp)g financial interest.

■ REFERENCES

(1) (a) Guerram, M.; Jiang, Z.-Z.; Zhang, L.-Y. Chin. J. Nat. Med. 2012, 10, 161−169. (b) Xu, H.; Lu, M.; Tian, X. Curr. Med. Chem. 2009, 16, 327−349. (c) Gordaliza, M.; Garcia, P. A.; del Corral, J. M.; Castro, M. A.; Gomez-zurita, M. A. Toxicon 2004, 44, 441−459. (d) Canel, C.; Moraes, R. M.; Dayan, F. E.; Ferreira, D. Phytochemistry 2000, 54, 115− 120. (e) Hartwell, J. L.; Schrecker, A. W. J. Am. Chem. Soc. 1951, 73, 2909−2916.

(2) (a) Mingoia, F.; Vitale, M.; Madec, D.; Prestat, G.; Poli, G. Tetrahedron Lett. 2008, 49, 760−763. (b) Wu, Y.; Zhang, H.; Zhao, Y.; Zhao, J.; Chen, J.; Li, L. Org. Lett. 2007, 9, 1199−1202. (c) Casey, M.; Keaveney, C. M. Chem. Commun. 2004, 184−185. (d) Medarde, M.; Rames, A. C.; Caballero, E.; López, J. L.; de Clairac, R. P-L.; Feliciano, A. S. Tetrahedron Lett. 1996, 37, 2663−2666. (e) Jones, D. W.; Thompson, A. M. J. Chem. Soc., Perkin Trans. 1 1993, 2541−2548. (f) Ward, R. S. Synthesis 1992, 719−730 and references therein.

(3) (a) Wu, Y.; Zhao, J.; Chen, J.; Pan, C.; Li, L.; Zhang, H. Org. Lett. 2009, 11, 597−600. (b) Stadler, D.; Bach, T. Angew. Chem., Int. Ed. 2008, 47, 7557−7559. (c) Reynolds, A. J.; Scott, A. J.; Turner, C. I.; Sherburn, M. S. J. Am. Chem. Soc. 2003, 125, 12108−12109. (d) Berkowitz, D. B.; Choi, S.; Maeng, J.-H. J. Org. Chem. 2000, 65, 847−860. (e) Bush, E. J.; Jones, D. W. J. Chem. Soc., Perkin Trans. 1 1996, 151−155. (f) Hadimani, S. B.; Tanpure, R. P.; Bhat, S. V. Tetrahedron Lett. 1996, 37, 4791−4794. (g) Speybroeck, R. V.; Guo, H.; Eycken, J. V.; Vandewalle, M. Tetrahedron 1991, 47, 4675−4682. (h) Andrews, R. C.; Teague, S. T.; Meyers, A. I. J. Am. Chem. Soc. 1988, 110, 7854−7858. (4) (a) Kondo, Y.; Suzuki, N.; Takahashi, M.; Kumamoto, T.; Masu, H.; Ishikawa, T. J. Org. Chem. 2012, 77, 7988−7999. (b) Khantikaew, I.; Takahashi, M.; Kumamoto, T.; Suzuki, N.; Ishikawa, T. Tetrahedron 2012, 68, 878−882. (c) Wannaporn, D.; Ishikawa, T.; Kawahata, M.; Yamaguchi, K. J. Org. Chem. 2006, 71, 6600−6603. (d) Wannaporn, D.; Ishikawa, T. J. Org. Chem. 2005, 70, 9399−9408. (e) Haga, T.; Ishikawa, T. Tetrahedron 2005, 61, 2857−2869. (f) Hada, K.; Watanabe, T.; Isobe, T.; Ishikawa, T. J. Am. Chem. Soc. 2001, 123, 7705−7706.

(5) (a) Hayashi, Y.; Kumamoto, T.; Nakanishi, W.; Kawahata, M.; Yamaguchi, K.; Ishikawa, T. Tetrahedron 2010, 66, 3836−3841. (b) Manaka, T.; Nagayama, S.-I.; Desadee, W.; Yajima, N.; Kumamoto, T.; Watanabe, T.; Ishikawa, T.; Kawahata, M.; Yamaguchi, K. Helv. Chim. Acta 2007, 90, 128−142.

(6) Yadav, J. S.; Subba, B. V.; Srinivasa, R.; Veerendhar, G.; Nagaiah, K. Tetrahedron Lett. 2001, 42, 8067−8070.

(7) Calculation of the electron density in the sesamol molecule using RHF/6-31G shows that the phenolic oxygen atom is the most electron rich position.

(8) Joseph, B.; Behard, A.; Lesur, B.; Guillaumet, G. ́ Synlett 2003, 1542−1544.

(9) (a) Beletskaya, I. P.; Cheprakov, A. V. Chem. Rev. 2000, 100, 3009− 3066. (b) Takahashi, H.; Fukami, T.; Kojima, H.; Yamakawa, T.; Sakamoto, T.; Nishimura, T.; Nakamura, M.; Yoshimizu, T.; Niiyama, K.; Ohtake, N.; Hayama, T. Tetrahedron 2005, 61, 3473−3481.

(10) Sonogashira, K.; Tohda, Y.; Hagiwara, N. Tetrahedron Lett. 1975, 50, 4467−4470.

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(11) Honda, T.; Ishikawa, F. Chem. Commun. 1999, 1065−1066.

(12) In the corresponding ethyl ester series, a larger coupling constant (*J* = 7.5 Hz) for the *trans* derivative^{12a} and a smaller constant (*J* = 4.7 Hz) for the cis derivative^{12b} were reported: (a) Pohmakotr, M.; Komutkul, T.; Tuchinda, P.; Prabpai, S.; Kongsaeree, P.; Reutrakul, V. Tetrahedron 2005, 61, 5311−5321. (b) Kaneko, T.; Wong, H. S. L. Eur. Pat. EP0224938, 1987-06-10.

 (13) Only ¹H NMR data for the structure assignment are given, due to its instability.